

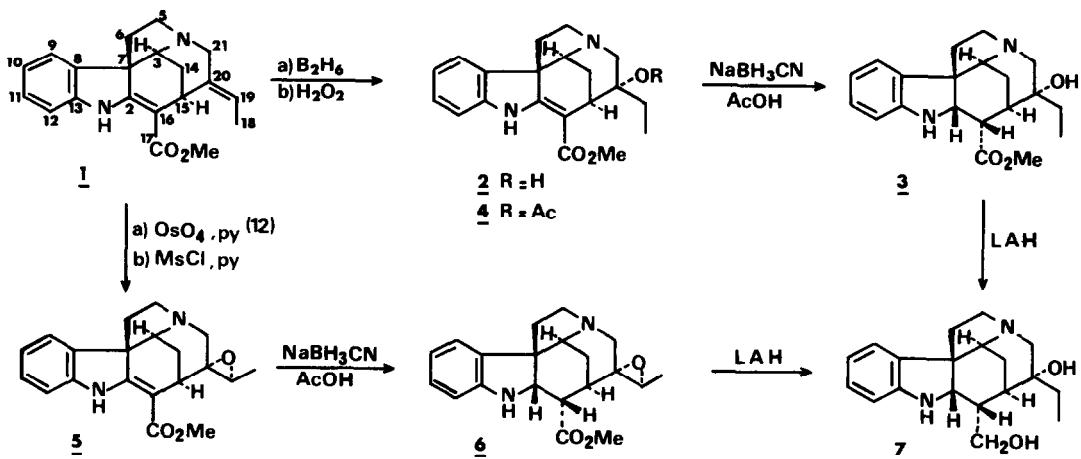
THE MARKOVNIKOFF HYDROBORATION OF A TRISUBSTITUTED OLEFIN

C.MIRAND, G.MASSIOT, L.LE MEN-OLIVIER and J.LEVY

Faculté de Pharmacie, ERA 319, CNRS, 51 rue Cognacq-Jay - 51096 REIMS Cédex FRANCE

Abstract : Hydroboration-oxidation of akuammicine 1 mainly gave the tertiary alcohol 2

Hydroboration-oxidation of double bonds is a well known anti-Markovnikoff process^{1,2}. In the course of applications of this method to the indole alkaloid field^{3,4,5*}, the following Markovnikoff hydration of a trisubstituted olefin was surprisingly observed :



Hydroboration-oxidation⁶ of akuammicine 1 gave 2⁷ (60%) and its 2,16-dihydroderivative 3 (10%), which was also prepared by $\text{NaBH}_3\text{CN}/\text{AcOH}$ ⁸ reduction of 2. The 20α -hydroxy group in 2 was suspected from its dehydration (TsOH , toluene) back to 1 (10%) and from its ^1H NMR (distorted methyl triplet) and mass spectra (m/z 283, $M^+ - \text{C}_3\text{H}_5\text{O}$). The ^{13}C NMR spectra of 2 and of its acetate 4^{9,10} showed all the carbon resonances at expected values. The shielding of C-14 in 2 was ascribed to a γ -interaction with the C-20 hydroxyl group¹¹, while C-15, C-19 and C-21 in 4 underwent the expected upfield shifts. Upon LAH reduction, 3 gave diol 7, which was identical with the compound prepared through successive reductions of epoxide 5¹² with $\text{NaBH}_3\text{CN}/\text{AcOH}$ (6) and then LAH, thus proving 2 to be 20-epi lochneridine.

* This work was initiated by the late Professor Jean LE MEN.

Formation of the tertiary alcohol 2 as the major product may be due to the firstly formed amine borane directing hydroboration on the carbon atom closer to nitrogen ; such an effect has previously been reported^{13,4} for cyclic disubstituted olefins. The electronegativity of the basic nitrogen should enhance such an effect¹⁴.

	<u>1</u>	<u>2</u>	<u>4</u>		<u>1</u>	<u>2</u>	<u>4</u>
C-2	167,3	168,8	168,6	C-14	30,9	25,9	25,9
C-3	61,8	60,7	60,5	C-15	29,8	37,3	33,5
C-5	56,8	54,2	54,2	C-16	100,8	101,6	99,8
C-6	46,2	43,0	43,1	C-17	167,3	171,2	170,4
C-7	57,5	56,6	57,0	C-18	12,8	7,0	7,0
C-8	136,6	135,9	135,9	C-19	119,9	33,3	27,4
C-9	120,4	121,2	121,3	C-20	138,9	70,4	63,6
C-10	120,3	119,9	119,8	C-21	56,2	54,3	51,7
C-11	127,2	127,6	127,7	COOCH ₃	50,4	51,0	51,1
C-12	108,9	109,7	109,8	OCOCH ₃			172,7
C-13	143,0	144,6	144,3	OCOCH ₃			22,2

Acknowledgements : We express our thanks to Pr J.CHUCHE and Dr J.BOUQUANT for measurements and interpretation of the ¹³C NMR spectra.

REFERENCES

- 1) H.C.Brown,"Organic Syntheses via Boranes",John Wiley and Sons,Inc.,New York NY (1975).
- 2) H.O.House,"Modern Synthetic Reactions",W.A.Benjamin,Inc.,Menlo Park,106 and ref. cited(1972).
- 3) G.Lukacs,M.de Bellefon,L.Le Men-Olivier,J.Lévy and J.Le Men,Tetrahedron Lett.,487 (1974).
- 4) C.Caron-Sigaut,L.Le Men-Olivier,G.Hugel,J.Lévy and J.Le Men,Tetrahedron 35,957 (1979).
- 5) C.Caron,L.Le Men-Olivier,M.Plat and J.Lévy,Heterocycles 16,645 (1981).
- 6) Diborane was generated *in situ* from NaBH₄-BF₃/Et₂O. With commercial BH₃/THF solution the proportion of 3 was reduced.
- 7) Compound 2:mp 205-206°C; (α)_D-706(MeOH);UV(MeOH) λ_{max} nm(log ϵ)236(4.02),296(3.99),328(4.15); IR(Nujol) ν cm⁻¹:3550,3350,1685,1585;MS 340(M⁺),322,283,226(100%),194,167,101,99;¹H NMR(CDCl₃,60MHz):δ0.97(3H,t,J=6.5Hz),3.77(3H,s),4.50(1H),6.70-7.50(4H,m),9.06(1H,s).
- 8) M.Döe de Maindreville,Thèse de Doctorat ès Sciences,Reims (1976).
- 9) Compound 4:mp 166-172°C; (α)_D-540(CHCl₃);UV(MeOH) λ_{max} nm 230,295,327;IR(Nujol) ν cm⁻¹: 3320,1710,1680,1580;MS 382(M⁺),322(100%),291,263,239,226,194,167,121;¹H NMR(CDCl₃,60MHz):δ0.88(3H,t,J=7.5Hz),2.10(3H,s),3.76(3H,s),6.60-7.50(4H,m),9.15(1H,s).
- 10) Under the acetylation conditions, the epimeric lochneridine (20B-OH) suffers dehydration : Y.Nakagawa,J.M.Wilson,H.Budzikiewicz and C.Djerassi,Chem.Ind. 1986 (1962).
- 11) J.B.Stothers, Carbon-13 NMR Spectroscopy,Academic Press,New York (1972).
- 12) B.J.Gilbert, A.P.Duarte, Y. Nakagawa, J.A.Joule, S.E.Florès, J.A.Brissolese, J.Campello, R.J.Owlen, E.C.Blossey, K.S.Brown and C.Djerassi, Tetrahedron 21,1141 (1965).
- 13) R.E.Lyle, K.R.Carle, C.R.Ellefson and C.K.Spicer, J.Org.Chem. 35,802 (1970).
- 14) H.C.Brown, M.K.Unni, J.Amer.Chem.Soc.,90,2902(1968) .